
A modelling assessment of acidification and recovery of European surface waters

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Abstract

The increase in emission of sulphur oxides and nitrogen (both oxidised and reduced forms) since the mid-1800s caused a severe decline in pH and ANC in acid-sensitive surface waters across Europe. Since *c.*1980, these emissions have declined and trends towards recovery from acidification have been widely observed in time-series of water chemistry data. In this paper, the MAGIC model was applied to 10 regions (the SMART model to one) in Europe to address the question of future recovery under the most recently agreed emission protocols (the 1999 Gothenburg Protocol). The models were calibrated using best available data and driven using S and N deposition sequences for Europe derived from EMEP data. The wide extent and the severity of water acidification in 1980 in many regions were illustrated by model simulations which showed significant deterioration in ANC away from the pre-acidification conditions. The simulations also captured the recovery to 2000 in response to the existing emission reductions. Predictions to 2016 indicated further significant recovery towards pre-acidification chemistry in all regions except Central England (S Pennines), S Alps, S Norway and S Sweden. In these areas it is clear that further emission reductions will be required and that the recovery of surface waters will take several decades as soils slowly replenish their depleted base cation pools. Chemical recovery may not, however, ensure biological recovery and further reductions may also be required to enable these waters to achieve the ‘good ecological status’ as required by the EU Water Framework Directive.

Keywords: Europe, acid-sensitive, waters, predictions, recovery, protocols.

Introduction

The impact of acidic deposition on surface waters in Europe is well documented and understood (Grennfelt *et al.*, 1995). This understanding has contributed to international legislation to reduce the emissions of sulphur (S) and nitrogen (N) which are responsible for the acidic deposition (Jenkins 1999; Bull *et al.* 2001). As a result of these agreements, the deposition of S has fallen dramatically across Europe and many surface waters have begun to

recover from acidification, as manifest by trends of decreased sulphate (SO₄), increased pH and increased acid neutralising capacity (ANC). The key issues that remain concern the timescale over which the recovery process will occur across Europe and the degree to which pre-acidification conditions will be reached. Both of these issues have significant implications for the review of the latest United Nations Economic Commission for Europe

(UNECE) Protocol under the Convention on Long-range Transboundary Air Pollution (CLRTAP), the 1999 Gothenburg or Multi-Pollutant Multi-Effect Protocol (UNECE 2002), and for new legislation within the European Union (EU) relating to standards for fresh water quality, the Water Framework Directive (WFD).

This paper makes a European scale assessment, using the output from dynamic acidification models, MAGIC and SMART, applied to acid-sensitive surface waters, to determine the extent and time to recovery and the implications with respect to achieving good ecological status in these ecosystems.

Background

The CLRTAP was established in 1979 and in the last 24 years has adopted four protocols to reduce the emission of S and N to the atmosphere (Jenkins 1999; Bull *et al.* 2001). The first S Protocol was signed in 1985 and committed signatories to reduce S emission by 30% by 1993 relative to 1980 levels. The second S (Oslo) Protocol was signed in 1994 and committed the signatories to a total emission reduction of c.80% of 1980 levels by 2010, but with different targets for each country. This was the first Protocol to be guided by the critical loads concept (Hettelingh *et al.*, 1995). The first N Protocol was signed in 1988 with an agreement to stabilise emissions at the 1987 level by 1994. Finally, the Multi-Pollutant, Multi-Effect (Gothenburg) Protocol was signed in 1999 with the aim of abating acidification, eutrophication and ground level ozone through emission ceilings for S, N and volatile organic compounds. If fully implemented by 2010, this would reduce S emissions across Europe by at least 63%, NO_x by 41% and NH_y by 17%, relative to 1990 levels (UNECE 2002), which are already substantially reduced from the peak levels of the late 1970s and 1980s as a result of the earlier agreements.

In parallel with these Protocols, the EU has developed and passed legislation to reduce S emission from large sources under the Large Combustion Plant Directive (NEGAP 2001). More recently, the EU has established the National Emission Ceilings Directive which limits national emissions of S, NO_x, NH₃ and Volatile Organic Compounds (Skeffington, 2002). By 2010, the overall effect of this legislation is expected to be a reduction in S emission to 85% of the 1980 level from the EU countries and to 74% of the 1980 level from the whole of Europe (Ferrier *et al.*, 2001).

The EU has also established a new structure for maintaining the quality of surface and groundwater, called the Water Framework Directive (WFD) (European Community 2000). The air pollution legislation and water

pollution legislation both will affect water quality in headwater lakes and streams that are sensitive to acidic deposition and where biological resources are degraded as a result of acidification. In this case, the WFD does not explicitly legislate to prevent pollution but expects measures to be adopted that have the effect of reducing the problem. The WFD has the stated purpose of establishing a framework for the protection of inland waters which prevents further deterioration and protects and enhances the status of aquatic ecosystems. The main objective is to achieve 'good' ecological status by 2016 relative to a 'reference condition'. 'Reference conditions' and 'good status' remain as yet undefined. With respect to acidification in small headwater lakes, a key indicator of ecological status is the presence or absence of a viable brown trout population.

The dose-response relationships between water chemistry and damage to aquatic organisms are well-documented with respect to freshwater acidification. Key chemical parameters related to biological effects are pH and concentrations of inorganic aluminium, calcium (Ca) and ANC (calculated as the sum of base cations minus strong acid anions (Baker *et al.*, 1987). This information has been used for calculating critical loads in Europe. The most widely used criterion is with respect to ANC, as both static and dynamic acidification models can be readily used to relate changes in acid deposition to changes in ANC in surface waters. Empirical relationships between ANC and fish population status have been developed for brown trout (*Salmo trutta L.*) and other species in Norway (Lien *et al.* 1992; Bulger *et al.* 1993; Lien *et al.* 1996). For brown trout in Norwegian lakes the probability of damage is < 10% if ANC > 20 µeq l⁻¹ and is 50% if ANC < zero (Fig. 1). For the more sensitive species Atlantic salmon (*Salmo salar L.*), the critical limit is ANC 30 µeq l⁻¹.

An acidification index for macro-invertebrates in Norway developed by Raddum and Fjellheim (1984) has been successfully applied in other regions in Europe (Raddum and Skjelkvåle 1995). For sites included in the ICP Waters database, the ANC limit suggested is 50 µeq l⁻¹ in central Europe, 30 µeq l⁻¹ in the central Alps and 20 µeq l⁻¹ in the British Isles and Scandinavia (Skjelkvåle *et al.* 2000).

In general, most organisms show a graded response to ANC rather than a step change and thus response curves can be fitted using logistic regressions. In this way, appropriate chemical targets can be defined to protect or establish a required biological status. For example, logistic regressions that describe the occurrence of the diatom *Achnanthes minutissima*; the macroinvertebrate *Baetis rhodani*; non-impooverished macroinvertebrate assemblages; number of mayfly species and the density of trout/salmonid populations in relation to surface water chemistry, in

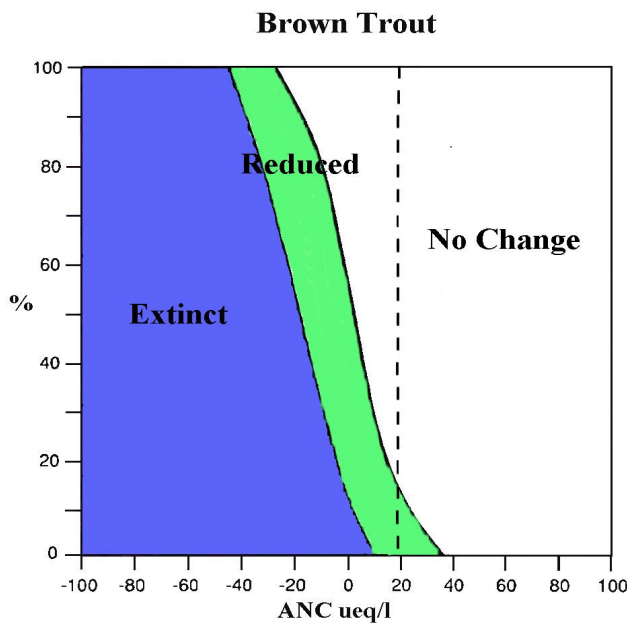


Fig. 1. Brown trout population status in relation to surface water ANC for 827 lakes in Norway (after Lien *et al.*, 1996). Note that the probability of damage to the fish population is < 10% at ANC 20 $\mu\text{eq l}^{-1}$ and < 50% at ANC zero.

particular ANC (Table 1), have been constructed from surveys undertaken in England, Scotland and Wales (Juggins *et al.*, 1995) and elsewhere in Europe (Skjelkvåle *et al.*, 2003). On the basis that critical loads for water considered in the existing protocols aim to protect brown trout, for evaluation purposes here, an ANC limit of 20 $\mu\text{eq l}^{-1}$ was used to indicate that the sites are capable of achieving the good ecological status as required by the WFD. With respect to other species, however, this ANC limit may not reflect good ecological status (Table 1).

Table 1. Predicted biological status, in terms of the diatom *A. minutissima* (probability of occurrence), the mayfly *B. rhodani* (probability of occurrence), number of mayfly species (number of taxa) and trout density (numbers per 100 m^2) for given surface water ANC concentration ($\mu\text{eq l}^{-1}$), after Juggins *et al.*, 1995.

ANC	Probability of occurrence		Number of taxa	Trout density
-100	0.01	0.03	0.02	5
-50	0.06	0.09	0.04	9
-20	0.17	0.18	0.06	12
0	0.30	0.28	0.08	14
20	0.48	0.39	1.1	17
50	0.74	0.59	1.8	23
100	0.95	0.85	4	37
200	0.99	0.99	19.4	94

Data sources

This analysis focused on acidified and acid-sensitive headwater lakes and streams in Europe as identified within the RECOVER:2010 and EMERGE projects (Table 2). The data used in this analysis are outputs from regional applications of the MAGIC model, except in Finland where the SMART model was used. MAGIC and SMART have been rigorously compared and found to be consistent in structure and output (Posch *et al.*, 2003a). In all cases, the model applications used the best available soil physical and chemical data and were calibrated to match present-day observed water chemistry. Regional estimates of S deposition were adjusted to match observed S concentrations in surface water on a site-by-site basis (Table 3). Sulphur adsorption and de-sorption were not considered important processes affecting the current output flux of S in the regions modelled. Consistent deposition sequences from 1860 to 2020 (Schopp *et al.*, 2003) were then used to scale the deposition through time.

These regional applications have been fully documented and discussed elsewhere: S Pennines, Wales and Galloway, SW Scotland (Helliwell *et al.*, 2003), N Scotland (Helliwell, *in press*), Finland (Posch *et al.*, 2003b), S Norway (Wright and Cosby, 2003a), Central Norway (Wright and Cosby, 2003b), S Alps (Rogora *et al.*, 2003), Tatra mountains (Kopacek *et al.*, 2003), Pyrenees (Camarero *et al.*, *in press*), N and S Sweden (Moldan, *et al.*, *in press*). The basis for this regional approach is that the physical and chemical catchment characteristics that mainly determine surface water chemistry are consistent for each group of lakes/streams, e.g. bedrock geology, atmospheric deposition, climate. The regions are, therefore, of different sizes and characterised by different numbers of lakes/streams (Table 2). An assumption behind this European assessment was that the sites modelled in each region comprise a sample characteristic of the total population of lakes/streams.

In the UK, the surface water chemistry surveys in the S Pennines (undertaken in 1998) and Galloway (mean of surveys undertaken in 1996 and 1997) included all reservoirs and lakes in the regions, respectively. The Pennine reservoirs are more than 100 years old and most are not currently in use as water supplies (Evans *et al.*, 2000). The water chemistry survey in Wales comprised 95 headwater streams sampled monthly during 1995 which enabled calculation of a reliable annual mean, but the sites were selected on the basis of accessibility and so are not necessarily representative of acid waters in the region. For N Scotland, the sites used are 30 headwater lochs, sampled during September-October 2000, that are representative of all lochs > 0.5 ha in the region.

Table 2. Regions modelled within the RECOVER:2010 and EMERGE projects. Sites are natural lakes except where indicated. More details on each are given in the reference indicated.

Region	No of Sites	Vegetation	Reference
Tatra Mountains	31	Alpine	Kopacek <i>et al.</i> , 2003
S Pennines, England	59	Moorland (reservoirs)	Helliwell <i>et al.</i> , 2003
S Alps	13	Forest (streams)	Rogora <i>et al.</i> , 2003
Wales	95	Forest, moorland (streams)	Helliwell <i>et al.</i> , 2003
Galloway, S Scotland	54	Forest, heathland	Helliwell <i>et al.</i> , 2003
Pyrenees	85	Alpine	Camarero <i>et al.</i> , in press
S Norway	60	Forest, heathland	Wright and Cosby, 2003
Central Norway	19	Forest heathland	Wright and Cosby, in press
S Sweden	35	Forest, heathland	Moldan <i>et al.</i> , in press
N Sweden	30	Forest, heathland	Moldan <i>et al.</i> , in press
N Scotland	30	Heathland	Helliwell <i>et al.</i> , in press
Finland	36	Forest, peatland	Posch <i>et al.</i> , 2003

The 60 lakes in southern Norway are small, acid-sensitive headwater lakes, selected from the 1986 and 1995 Norwegian lakes surveys (Henriksen *et al.*, 1988, Henriksen *et al.*, 1997, Skjelkvåle *et al.*, 2001). The lakes are characteristic of the majority of the several thousand lakes in the region. The 19 lakes in central Norway were selected for the EMERGE project (www.mountain-lakes.org) on the basis of the availability of biological data from previous investigations. These lakes are characteristic of acid-sensitive lakes in the region. Southern Norway received moderate levels (on a European scale) of acid deposition and the lakes in the region are highly acidified. Central Norway receives very low levels of acid deposition and the lakes in the region are only slightly acidified and thus serve as unpolluted references.

In the Tatra Mountains, 31 lakes were selected in the western part and central massive on the basis of a previous survey of water quality (Kopáček *et al.*, 2000); the eastern part has no lakes. Of the total lake population in the district (138 perennial lakes > 0.01 ha), the selected 31 lakes were both regionally and morphologically representative for the alpine zone. The selected lakes included both seepage and drainage lakes and are all of glacial origin.

In the S Alps, only a small number of sites (three streams, ten alpine lakes) were used. These were chosen to be characteristic of the whole alpine and sub-alpine area in North-Western Italy. The sites have different sensitivities to acidification and reflect the geological complexity of the area, which is from igneous to carbonaceous rocks (Rogora *et al.*, 2001).

In Finland, the sites were a subset of 36 of the 163 lakes in the Finnish acidification monitoring lake network

(RMLA; Forsius *et al.*, 2003). The lakes are mostly small (median area = 10 ha), acid-sensitive, head water or seepage lakes and their catchments are forested and relatively undisturbed. These lakes were originally chosen subjectively by expert judgement for use in national acidification studies. The 36 lakes were sampled in 1987 and then annually since 1990.

In Sweden, 65 intensively monitored lakes were modelled in this study. These are part of a national survey of water quality (Riksinventeringen, RI) started in 1972 (Bernes 1986) and repeated five times (Wilander *et al.*, 1998). The monitoring programme was designed to represent all types of lakes that commonly occur in Sweden, especially with respect to TOC concentrations, acidity, size and geographical position (SNV, 1993).

In the Pyrenees, 85 lakes were sampled during the summer 2000. These were selected on the basis of altitudinal and lithologic distributions of the entire population of lakes in the Pyrenean range (Camarero *et al.*, 2003). Typically, the lakes in the Pyrenees are located above the tree line, are relatively small and have short water renewal times. The majority of lakes are on acid sensitive catchments composed mainly of granites and granodiorites, but many lakes are located on carbonated, less sensitive rocks.

For predictions to 2016, the reduction in deposition of S, NO_x and NH₃ have been calculated (Table 3) using the RAINS model on the 150 × 150 km EMEP grid (Schopp *et al.*, 2003). These assume that the agreed reductions under the UN-ECE Protocols and EU legislation will be implemented and comprise the 'current legislation' scenario (CLE) for 2010. The present day S deposition at each site in each region derived from the observed surface water

Table 3. Regional mean non-marine S deposition flux used in the model simulations. Fluxes are in $\text{meq m}^{-2} \text{yr}^{-1}$ and percent reduction (shown in parentheses) is relative to 1980. 1980 and 2016 estimates are based on the deposition patterns derived for the relevant EMEP grid square (Schopp *et al.*, 2003).

Region	$\times \text{SO}_4$ deposition flux		(meq m ² yr)
	1980	2000	
Tatras	146	63 (57%)	51 (65%)
Pennines	328	105 (68%)	46 (86%)
S.Alps	95	63 (34%)	29 (70%)
Wales	73	22 (70%)	11 (85%)
Galloway	228	73 (68%)	40 (82%)
Pyrenees	66	34 (49%)	23 (65%)
S.Norway	157	55 (65%)	36 (77%)
Central Norway	49	17 (65%)	11 (78%)
S. Sweden	78	35 (55%)	21 (73%)
N. Sweden	38	17 (55%)	13 (66%)
N.Scotland	52	30 (42%)	18 (65%)
Finland	66	18 (73%)	15 (77%)

concentration was then reduced by the same percentage to 2010 (Table 3). Beyond 2010, deposition was assumed to remain constant.

Results

Background pre-acidification (1860) surface water ANC concentrations were in almost all cases above zero and predominantly above $20 \mu\text{eq l}^{-1}$ (Fig. 2). With the rapid increase in emission and deposition of S and N from 1860 to 1980, the occurrence of acidic surface waters, indicated by ANC less than zero, became widespread in sensitive regions across the whole of Europe (Fig. 3). The dominant cause of this decline in ANC was the increase in SO_4 concentration. By 2000, the effect of international agreements and European legislation has led to a reduction in SO_2 emission of *c.*55% across Europe and this has led to a similar reduction in S deposition (NEGTA, 2001). These deposition reductions have been successful in promoting recovery from surface water acidification across Europe (Fig. 4). ANC in all regions was simulated to have increased over the last 20 years, a result in agreement with empirical evidence on trends in surface water recovery (Evans *et al.*, 2001).

The model predictions to 2016 under the CLE scenario (Table 3) showed the potential benefit of the currently proposed reductions (Fig. 5). ANC concentrations in sensitive surface waters across most of Europe were predicted to recover and most waters reached levels above

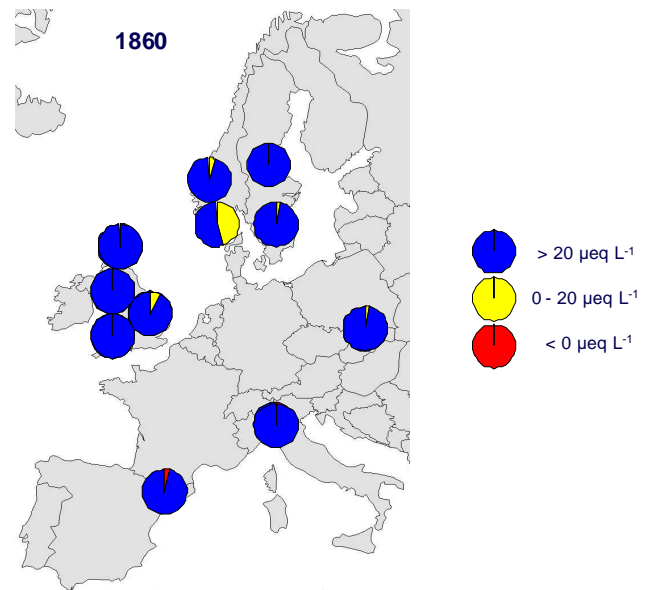


Fig. 2. The 1860 model reconstructed surface water ANC concentration for each region expressed in 3 ANC classes. For Finland, the SMART model is used and this is initialised in 1960 and so no data are available.

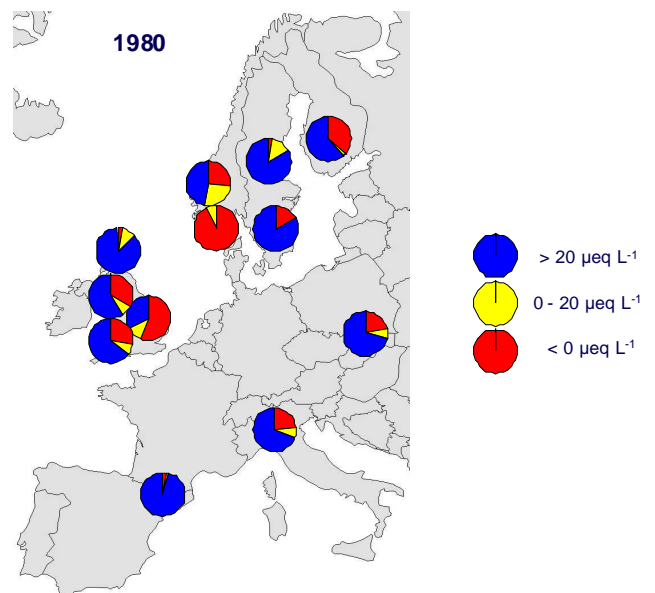


Fig. 3. The 1980 (worst condition) model reconstructed surface water ANC concentration for each region expressed in 3 ANC classes.

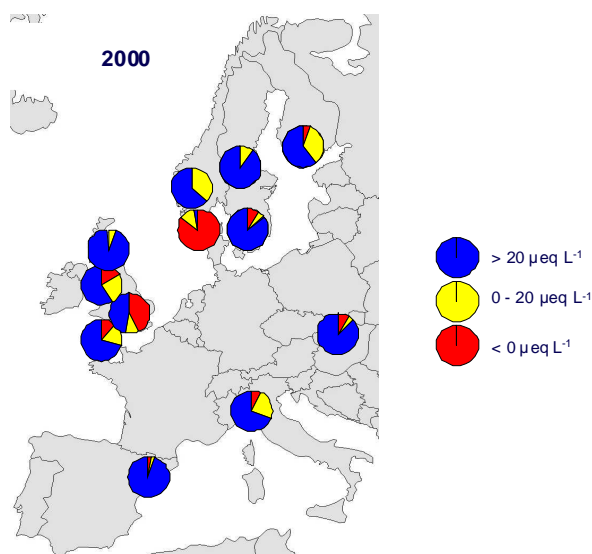


Fig. 4. The 2000 observed surface water ANC concentration for each region expressed in 3 ANC classes.

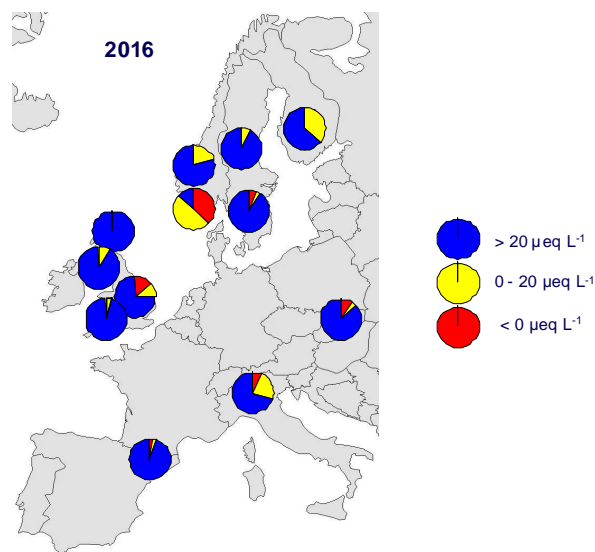


Fig. 5. The 2016 model predicted surface water ANC concentration for each region expressed in 3 ANC classes.

$20 \mu\text{eq l}^{-1}$. This does not imply, however, that these systems have returned to their background conditions with respect to either chemical or biological characteristics.

At the Europe scale distinct patterns emerged from the model simulations through time. In the UK, the simulations indicated pre-acidification conditions with ANC generally above $20 \mu\text{eq l}^{-1}$. By 1980, after many decades of significantly high levels of S deposition, severe acidification occurred across the whole country. By 2000, the signs of recovery were apparent and by 2016 it is only in the S Pennines that more than 10% of surface waters were predicted to be below zero ANC. The S Pennine region lies very close to major cities and local S deposition is extremely high (Table 3). Nevertheless, the deposition reductions by 2010 represent a very significant reduction from 1980 levels (65–86%, Table 3) and the simulations indicated that the recovery of surface water ANC in the Pennines will be relatively rapid (a few decades from present).

In Scandinavia, acidification equally dramatic as that in UK situation was simulated between 1860 and 2000. Despite generally lower peak S deposition fluxes in 1980 but similar magnitude of deposition reduction to 2010 (66–77%, Table 3), however, the predicted recovery by 2016 was not large. A significant proportion of surface waters were predicted to have ANC less than $20 \mu\text{eq l}^{-1}$, and in S Norway and S Sweden ANC less than zero will still be widespread. This implies the need for additional deposition reductions and recovery may take much longer (several decades).

In the S Alps and the Tatra in central Europe the situation is very similar to Scandinavia; despite *c.* 65% reductions in S deposition from 1980 levels by 2010, significant

acidification was predicted to remain in 2016. In the Pyrenees, acidification has not occurred to a significant degree and ANC was predicted to be back at *c.* 1860 levels by 2010.

Discussion

The ANC simulations for 1860 (pre-acidification condition) indicated that chemical conditions across Europe were generally suitable for supporting brown trout — in nearly all cases ANC was greater than $20 \mu\text{eq l}^{-1}$. ANC $20 \mu\text{eq l}^{-1}$ can be used as a threshold for defining ‘good status’ with respect to the WFD as well as a recovery target in terms of CLRTAP. The problem of generalising this as an indicator of a ‘reference condition’ against which current status and future predictions can be assessed is that some waters had background predicted ANC below $20 \mu\text{eq l}^{-1}$ in particular in the S Pennines, the Pyrenees, S Norway and the Tatra Mountains (Fig. 2). Low ANC may be due to the dilute nature of the water as a result of low rates of weathering and thin soils or from high concentrations of SO_4 derived from mineral weathering. The latter is almost certainly the cause in the Pyrenees where a significant contribution from weathering to SO_4 concentrations has been identified at some sites (Catalan *et al.*, 1993). In S Norway and the Tatra Mountains, thin and organic-rich soils combined with low base cation weathering rates are the likely cause of the low background ANC levels. In the S Pennines, SO_4 from mineral weathering probably contributed to low background ANC levels.

A further problem in considering the simulated

background condition as an indicator of a reference condition is that the pre-acidification biological status of many of these sites with respect to brown trout populations and other key indicator organisms is not known. The simulated water chemistry in 1860, however, should have been sufficient for brown trout although other hydromorphological, physical or chemical factors may have prevented them from being present.

By 1980, deposition of S reached a maximum and both observed and simulated surface water acidification was widespread and severe across all of Europe (Fig. 3). The simulated acidification agrees with reports of widespread acidification problems at that time in the UK, Scandinavia and Central Europe (e.g. Drablos and Tollan, 1980). In regions of Europe with substantial S deposition, more than 25% of sensitive waters had ANC below zero and would have been unlikely to support brown trout. The substantial changes from background levels imply, therefore, that fish populations had been damaged across wide areas. The extent of the problem around 1980 prompted the establishment of the CLRTAP and the initial negotiations for reductions in the emission of acidifying pollutants. There exists little or no monitoring data to verify the modelled decreases in ANC before 1980, but systematic observations that began in the late 1970s indicate severe acidification (e.g. Overrein *et al.*, 1980). In addition, there exists a well established body of evidence from sediment core diatom reconstructions of surface water pH that indicate a severe acidification at many sites in these regions (e.g. Battarbee *et al.*, 1990).

Between 1980 and 2000 there have been significant observed improvements in ANC of acidified surface waters in Europe (Stoddard *et al.*, 1999; Evans *et al.*, 2001) and these improvements were simulated by the model applications (Fig. 4). The improvements are largely in response to major decreases in S deposition in all regions (Table 3) and demonstrate the success of the policies and agreements already in place for reducing emission of acidifying compounds. It is also widely reported (Evans *et al.*, 2001; Stoddard *et al.*, 1999) that recovery in pH and Al have not always occurred in conjunction with the improved ANC and this may have consequences for the timescale of biological recovery. Significant acidity problems remain, however, in parts of the UK, Norway, Finland and the S Alps, where many waters have ANC below 20 $\mu\text{eq l}^{-1}$. The situation in these areas is significantly different from the reconstructed background or reference condition and so imply a significant 'departure' from 'good status' with respect to the WFD. Nevertheless, the rapid recovery from the 1980 worst case levels emphasises the benefits that may be achieved by making further reductions to 2010 under the existing agreements.

The model predictions for 2016, the key year with respect to achieving good status under the WFD, indicated that surface water acidification will be much less of a problem across most of Europe (Fig. 5). This assumes, of course, that the emission reductions agreed under the Gothenburg Protocol and existing and proposed EU legislation will actually be implemented by 2010. In Norway, Sweden, Finland, the S Alps and the S Pennines model predictions indicated that a significant percentage of waters will still have ANC less than 20 $\mu\text{eq l}^{-1}$ and waters in these areas remain significantly different from their simulated pre-acidification condition. In S Norway, there remains the issue that many lakes have background ANC less than 20 $\mu\text{eq l}^{-1}$ and so it is not relevant to aim for improvement beyond that level. In general, however, there is clear need for further emission reductions to reduce deposition to these areas.

Sulphur is the the key driver of the acidification and now recovery of surface waters across Europe. Nitrogen is not considered here as it plays a secondary role in most regions, although as SO_4 concentrations decrease, nitrate will become increasingly important. The speed of change in surface water ANC, both in terms of acidification and recovery lies not only in the magnitude of change in S deposition but in the soil characteristics and weathering rates in the regions. The key process in catchments for neutralising acid inputs is the exchange of base cations from soils. This store of exchangeable cations (the product of cation exchange capacity, the base saturation percentage and the bulk density of the soil) is in turn mainly replenished by base cations supplied from weathering of soil minerals. The pool of exchangeable base cations can only recover when the S deposition decreases to below the critical load.

In soils with low rates of weathering of base cations, the base cation pool has been depleted as S deposition increased. The model simulations showed reduced % base saturation between 1860 and 1980 (Table 4). Between 1980 and 2016, the model results suggest that the large deposition reductions in all regions will be insufficient to increase % base saturation and in most regions the soil will continue to acidify. It is possible that the base saturation will recover over a much longer time period (several decades from present) at the lower level of S deposition but it is more likely that the restoration of the original base status of catchment soils will require further emission reductions and will take many decades. For surface waters, therefore, even after the S deposition flux is reduced, ANC will remain below pre-acidification levels until the soil base cation pool is replenished (cf. Reuss *et al.*, 1987). In areas where the weathering rate is higher, the rate of recovery can be equally rapid as the rate of acidification, as exemplified in the S Pennines.

Table 4. Model simulated regional median and 25 percentile values of soil % base saturation in 1860, 1980 and 2016.

Region	1860			1980			2016		
	(25%ile)	Median	(75%ile)	(25%ile)	Median	(75%ile)	(25%ile)	Median	(75%ile)
Tatras	(14)	16	(18.5)	(10.5)	11.5	(12.2)	(10.5)	13.2	(13.7)
Pennines	(15.7)	18	(21)	(11.2)	11.7	(12.2)	(9.7)	10.2	(11)
S.Alps	(14)	22	(24)	(10.2)	18.2	(20.5)	(7.5)	9.5	(16)
Wales	(10.7)	14	(22.2)	(7.2)	10.5	(15.7)	(6.7)	9.5	(15)
Galloway	(8.7)	12	(17)	(5.2)	6.5	(9.5)	(4.2)	5.7	(8)
Pyrenees	(31.4)	43.4	(51.6)	(21.9)	34.5	(43.4)	(24.3)	38.1	(44.8)
S.Norway	(29)	35	(46)	(26)	27	(36)	(26)	29	(37)
C Norway	(20.5)	21.5	(22.5)	(18.7)	20	(20.2)	(18.7)	19.7	(20)
S. Sweden	(30)	36	(40)	(21)	23	(29)	(18)	21	(27)
N. Sweden	(29)	34	(39)	(24)	28	(36)	(21)	26	(35)
N.Scotland	(3.7)	6.5	(11.7)	(2.7)	5	(10.7)	(3.2)	5.5	(11.2)
Finland	(24)	39	(67)	(21)	37	(67)	(20)	36	(62)

If the acidification status of all European acid-sensitive surface waters is to be returned to levels which are capable of maintaining brown trout populations or another relevant indicator of similar sensitivity (= good ecological status), further reductions of S (and potentially N) emission will be required. These must encompass consideration of catchment soil characteristics, weathering rates, historical, present and future deposition in conjunction with a specified chemistry (ANC) target to be reached within a given timescale. Dynamic models such as MAGIC and SMART are currently capable of providing this assessment and model outputs are available for use in assessing the potential recovery in response to further emission reduction scenarios and in defining new emission reduction targets (Jenkins *et al.*, 2002). Further assessments must also be capable of including the potential impact of other environmental drivers, such as climate change, which may have an impact on the recovery from acidification over the same time period (Skjelkvåle *et al.*, 2002).

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